Physics 370

Thermal Physics

Fall 2022

Instructor: Daniel Robb Class Times: MWF 1:10-2:20 (Trexler 272)
Office: Massengill 243 Office Hrs: MWF 11:00-11:30 and 3:00-3:30

T/Th 1:30-3:00

Email: robb@roanoke.edu (15-min Zoom appts at calendly.com/daniel_robb)

Phone: 375-5250

Course Description:

Thermal behavior of systems; energy and entropy; equations of state; changes of phase; elements of continuum and statistical approaches

Textbook:

• *Thermal Physics* by Ralph Baierlein. Cambridge University Press, 1999. (Available in paperback) ISBN-13: 978-0521658386.

Purpose of the Course:

What do automobile engines, cloud formations and rubber bands have in common? The behavior of each of these systems depends on the flow of matter and energy among the constituent elements of these systems and the surrounding environment. Whereas mechanics, electricity and magnetism, and quantum mechanics describe the behavior of individual particles under the influence of forces, thermal physics attempts to explain and predict the behavior of large collections of particles.

"A theory is the more impressive the greater the simplicity of its premises, the more different kinds of things it relates, and the more extended its area of applicability. Hence the deep impression that classical thermodynamics made upon me. It is the only physical theory of universal content concerning which I am convinced that, within the framework of the applicability of its basic concepts, it will never be overthrown" – Albert Einstein.

This course will give you a practical understanding of how to use classical thermodynamics, as well as an appreciation for its wide range of applicability. We will work hard to understand the link between the mathematical formalism of statistical mechanics, which is rooted in the microscopic properties of systems, and the macroscopic properties (temperature, pressure, volume, etc.) described by classical thermodynamics. Finally, we will look at an application of thermal physics – the modeling of the Earth's climate – gaining an understanding of the types of models used, as well as separating fact from fiction in the current "climate debate".

Specific Goals of the Course:

- 1. Acquire the ability to apply classical thermodynamics to physical systems, and understand the Three Laws of Thermodynamics
- 2. Understand the link between statistical mechanics and thermodynamics, and gain a beginning proficiency in "stat mech"
- 3. Study an issue at the intersection of science and politics (climate change)

Feedback and Evaluation:

I will assign numerical grades to all your work. I *may* curve your final grades (upward), but otherwise you can expect to receive an "A" for 90-100, a "B" for 80-89, etc. I will assign +/- to final grades by examining the distribution of grades and my perception of your effort. These are the categories and percentages:

<u>Problem sets</u>: 35% (5 @ 7 % each) <u>Tests</u>: 30% (2 @ 15 % each)

Final exam: 20% Writing Assignment: 10%

Participation: 5%

<u>Problem sets</u>: I encourage you to discuss problems with other students, but the work you turn in should be your own (i.e., don't copy work from another student, or allow another student to copy your work.) See below for the policy on late work.

<u>Tests</u> during the semester will be given in our classroom during class periods. Each test will consist of several conceptual questions requiring written responses, and several calculation problems. The first test will cover the first two course units, i.e., classical thermodynamics and kinetic/transport theory. The second test will cover the third course unit on statistical mechanics. Note that *you will be given all necessary formulas on each test*.

The <u>final exam</u> will be comprehensive, including all four course units. It will also include conceptual questions and calculation problems.

The <u>writing assignment</u> will concern our short unit on theories of Earth's climate. You will summarize and critique article from the mainstream press in light of our study of climate modeling and currently available climate data. No collaboration is allowed on this assignment (except with me), and you must cite any sources you have used in footnotes. You will be required to turn in a rough draft (worth 1/3 of the grade), and then to revise your draft based on my comments (2/3 of the grade). See below for the policy on late work.

Your <u>participation grade</u> is based on your reflections on (at least) <u>two</u> MCSP Colloquium Series talks, as well as on your class attendance.

MCSP Colloquium Series:

This semester a series of talks will be offered which appeal to a broad range of interests related to math, computer science and physics. Members of this class are invited to attend all of these talks; however, participation in <u>at least two</u> of these sessions is mandatory. Within **one week** of attending a talk you must submit (via Inquire) a one-page single-spaced paper. This paper should not only include a summary of the main content of the talk, but also a personal contemplation of the experience.

Policy on Late Work:

I will grade an assignment with a 10% lateness deduction for each successive school day they are late (schooldays are M-F; days end at 5:00 PM). As a result, after one week assignments receive a 50% deduction. After two weeks, assignments receive a 100% deduction; that is, no assignment will be accepted if more than two weeks late. I will consider adjusting the late policy for an assignment only under extreme circumstances.

Academic Integrity:

The College's academic integrity policies will be enforced. Although you are encouraged to work in groups on your problem sets, all work turned in for a grade must be your own. See above for guidelines on the written assignment. Please familiarize yourself with the College's academic integrity policies. Be aware that I am contractually obligated to report students if I suspect that they have engaged in academic dishonesty. Lastly, unless otherwise directed, cell phones should be silenced and out of sight during all class periods.

Make-up Tests:

Make-up tests will not be given. If you miss a test, and have an official college excuse for that absence, then your final exam grade will count for the missed test.

Attendance Policy:

If you have a temperature of 100.4 or higher or other coronavirus symptoms, don't come to class. Call Health Services IMMEDIATELY. Do not come to class or go to any public area on campus. In order for your absence to be excused, you must give Health Services permission to notify me that you have consulted them about coronavirus symptoms. If Health Services informs you that you should isolate and not attend class for multiple days, contact me so that we can make a plan to keep you current in the course. All absences caused by consultation with Health Services about coronavirus symptoms or isolation ordered by Health Services will be excused, but you will need to do the work and graded assignments even if we extend a deadline for you.

The following is the course policy for all other absences. You are expected to attend every class. Attendance is checked at each class meeting. If you are going to be absent from class for a valid (excused) reason, I must be notified in advance. Your fourth and each additional unexcused absence will result in a 2-point deduction in your final course grade. Furthermore, you are accountable for all work missed because of any absence. I will provide class materials for a missed class, but will not re-teach a missed class during office hours.

Mask Policy

The College is starting the term without a specific mask mandate. I am not requiring masks in class, but you may wear certainly wear a mask in class if you choose to.

Use of Electronic Devices:

In class, you may use personal laptops, but only for course-related purposes. All other electronic devices must be turned off. On tests, you may use a scientific calculator; all other electronic devices must be turned off and out of sight. Violation of this policy on tests will be treated as a violation of the Academic Integrity policy.

Accessible Education Services:

Accessible Education Services (AES) is located in the Goode-Pasfield Center for Learning and Teaching in Fintel Library. AES provides reasonable accommodations to students with documented disabilities. To register for services, students must self-identify to AES, complete the registration process, and provide current documentation of a disability along with recommendations from the qualified specialist. Please contact Becky Harman, Assistant Director of Academic Services for Accessible Education, at 540-375-2247 or by e-mail at aes@roanoke.edu to schedule an appointment. If you have registered with AES in the past and would like to receive academic accommodations for this semester, please contact Becky Harman at your earliest convenience to schedule an appointment and/or obtain your accommodation letter for the current semester. If you are on record with AES as having academic or physical needs requiring accommodations, please contact me as soon as possible. We need to discuss your accommodations before they can be implemented. Also, please note that arrangements for extended time on exams and/or testing in a semi-private setting must be made at least one week *before every exam*.

The Writing Center @ Roanoke College, located on the Lower Level of Fintel Library (Room 15), offers free tutorials focused on writing projects and oral presentations for students working in any field. Writers and presenters at all levels of competence may visit the Writing Center at any point in their process—including brainstorming, drafting, organizing, editing, or polishing presentation skills—to talk with trained peer tutors in informal, one-on-one sessions. The Writing Center is open Sunday through Thursday from 4 to 9 PM. Simply stop in, or schedule an appointment at www.roanoke.edu/writingcenter. Questions? Email writingcenter@roanoke.edu or call 540-375-4949.

<u>#</u>	<u>Date</u>	<u>Topic</u>	Reading	<u>Due</u>
		UNIT 1: Classical thermodynamics		
1	Aug. 31	Introduction and preview		
2	Sept. 2	Heat, temperature and the 1st Law	1.1-1.4	
3	5	Adiabatic processes	1.5-1.7	
4	7	Multiplicity and the 2 nd Law	2.1-2.3	
5	9	Entropy I	2.4-2.8	
6	12	The Carnot cycle	3.1-3.4	
7	14	Reversibility and real engines	3.5-3.7	
8	16	Problem set workshop		
		UNIT 2: Transport theory		
9	19	Random walks	15.1-15.2	PS 1
10	21	Momentum transport and viscosity	15.3-15.4	
11	23	Thermal transport and diffusion	15.5-15.6	
12	26	Climate modeling I	6.4, Handouts	
13	28	Climate modeling II	Handouts	PS 2
14	30	Review and catch-up		
15	Oct. 3	TEST 1		
		UNIT 3: Statistical mechanics		
16	5	Density of states	4.1	
17	7	General definition of temperature	4.2-4.4	Paper draft
18	10	Thermal probabilities	5.1-5.3	·
19	12	The partition function	5.4-5.5	
20	14	The canonical distribution	5.6-5.8	Paper final, PS 3
	15-23	FALL BREAK		
21	24	Chemical potential I	7.1	
22	26	Chemical potential II	7.2-7.5	
23	28	Ideal gas: quantum treatment	8.1-8.3	
24	31	Ideal gas: classical limits	8.4	
25	Nov. 2	Free energy	10.1-10.3	
26	4	No class		
27	7	"Minimize the free energy"	10.4-10.8	
28	9	Chemical equilibrium	11.1-11.2	
29	11	Chemical equilibrium II	11.3	
30	14	Classical stat mech	13.1-13.2	PS 4
31	16	Equipartition theorem	13.3	
32	18	TEST 2		
		UNIT 4: Phase Transitions and the 3 rd Law		
33	21	Phases of matter I	12.1-12.3	
	23-27	THANKSGIVING BREAK		
34	28	Phases of matter II	12.4-12.5	
35	30	Fermions at low temperature I	9.1	
36	Dec. 2	Fermions at low temperature II	9.1	
37	5	Prelude to the 3 rd Law	14.1-14.3	PS 5
38	7	The 3 rd Law	14.4-14.5	
39	9	Review and catch-up		
		FINAL: Wed, Dec 14, 2:00-5:00pm		

Note: You should expect to spend a combined total of 12 hours per week on lecture, homework, and reading for PHYS 370.