

Instructor: Daniel Robb

Class Times: MWF 1:10-2:20 (OLIN 230)

Office: Massengill 243

Office Hrs: T 9-11, W 3-4, via Zoom:

Email: robb@roanoke.edu

Meeting ID 483 818 7781, Passcode 2583

Phone: 375-5250

Course Description:

Thermal behavior of systems; energy and entropy; equations of state; changes of phase; elements of continuum and statistical approaches

Textbook:

• Thermal Physics by Ralph Baierlein. Cambridge University Press, 1999. (Available in paperback)
ISBN-13: 978-0521658386.

Purpose of the Course:

What do automobile engines, cloud formations and rubber bands have in common? The behavior of each of these systems depends on the flow of matter and energy among the constituent elements of these systems and the surrounding environment. Whereas mechanics, electricity and magnetism, and quantum mechanics describe the behavior of individual particles under the influence of forces, thermal physics attempts to explain and predict the behavior of large collections of particles.

“A theory is the more impressive the greater the simplicity of its premises, the more different kinds of things it relates, and the more extended its area of applicability. Hence the deep impression that classical thermodynamics made upon me. It is the only physical theory of universal content concerning which I am convinced that, within the framework of the applicability of its basic concepts, it will never be overthrown” – Albert Einstein.

This course will give you a practical understanding of how to use classical thermodynamics, as well as an appreciation for its wide range of applicability. We will work hard to understand the link between the mathematical formalism of statistical mechanics, which is rooted in the microscopic properties of systems, and the macroscopic properties (temperature, pressure, volume, etc.) described by classical thermodynamics. Finally, we will look at an application of thermal physics – the modeling of the Earth’s climate – gaining an understanding of the types of models used, as well as separating fact from fiction in the current “climate debate”.

Specific Goals of the Course:

1. Acquire the ability to apply classical thermodynamics to physical systems, and understand the Three Laws of Thermodynamics
2. Understand the link between statistical mechanics and thermodynamics, and gain a beginning proficiency in “stat mech”

3. Study an issue at the intersection of science and politics (climate change)

Feedback and Evaluation:

I will assign numerical grades to all your work. I may curve your final grades (upward), but otherwise you can expect to receive an “A” for 90-100, a “B” for 80-89, etc. I will assign +/- to your final grades by examining the distribution of grades and taking into account my perception of your effort in the course. These are the categories and percentages that will be used:

<u>Problem sets:</u>	35% (5 @ 7 % each)	<u>Tests:</u>	30% (2 @ 15 % each)
<u>Final exam:</u>	20%	<u>Writing Assignment:</u>	10%
<u>Participation:</u>	5%		

Problem sets: I encourage you to discuss problems with other students, but the work you turn in should be your own (i.e., don't copy work from another student, or allow another student to copy your work.) See the next page for the policy on late work.

Tests during the semester will be given in our classroom during class periods. Each test will consist of several conceptual questions requiring written responses, and several calculation problems. The first test will cover the first two course units, i.e., classical thermodynamics and kinetic/transport theory. The second test will cover the third course unit on statistical mechanics. Note that you will be given all necessary formulas on each test.

The final exam will be comprehensive, including all four course units. It will also include conceptual questions and calculation problems.

The writing assignment will concern our short unit on theories of the Earth's climate. You will be required to summarize and critique article(s) from the mainstream press in light of our study of climate modeling and currently available climate data. No collaboration will be allowed on this assignment (except with me), and you must cite any sources you have used in footnotes. You will be required to turn in a rough draft (worth 1/3 of the grade), and then to revise your draft based on my comments (2/3 of the grade). See the next page for the policy on late work.

Your participation grade is based on your reflections on (at least) two MCSP Colloquium Series talks, as well as on your class attendance.

MCSP Colloquium Series:

This semester a series of virtual talks will be offered which appeal to a broad range of interests related to math, computer science and physics. Members of this class are invited to attend all of these talks; however, participation in at least two of these sessions is mandatory. Within one week of attending a virtual talk you must submit (via Inquire) a one-page single-spaced paper. This paper should not only include a summary of the main content of the talk, but also a personal contemplation of the experience

Policy on Late Work:

I will grade an assignment with a 10% lateness deduction for each successive school day they are late (school days are M-F; days end at 5:00 PM). As a result, after one week assignments receive a 50% deduction. After two weeks, assignments receive a 100% deduction; that is, no assignment will be accepted if more than two weeks late. Under extreme circumstances only will I consider adjusting the late policy for an assignment.

Make-up Tests:

Make-up tests will not be given. If you miss a test, and have an official college excuse for that absence, then your final exam grade will count for the missed test.

Attendance Policy:

If you have a temperature of 100.4 or higher or other coronavirus symptoms, call Health Services IMMEDIATELY if living on campus, or call your family doctor if living at home. Do keep up with all readings, assignments, and deadlines. If Health Services or your doctor informs you that you should isolate for multiple days or weeks, given that the class is online, you may still attend class, or watch cloud recordings of the classes if you prefer. All absences caused by consultation with Health Services or your family doctor about coronavirus symptoms or isolation will be excused.

The following is the course policy for all other absences. You are expected to attend every class. Attendance is checked at each class meeting, and you must be in class to participate in the in-class activities which form part of the class participation grade. If you are going to be absent from class for a valid (excused) reason, I must be notified in advance. Your fourth and each additional unexcused absence will result in a 2-point deduction in your final course grade. Furthermore, you are accountable for all work missed because of any absence. I will provide class materials for a missed class, but will not re-teach a missed class during office hours.

Face Covering Policy:

Face coverings/masks must be worn over the mouth and nose by all students and instructors in classrooms and hallways of academic buildings. By wearing face coverings, we protect our college community and its most vulnerable members.

Technology challenges:

You will need internet connectivity. If you have technology challenges, I need you to email me so that we can discuss how you can keep up.

Academic Integrity:

The College's academic integrity policies will be enforced. Although you are encouraged to work in groups on your problem sets, all work turned in for a grade must be your own. See the previous page for guidelines on the written assignment. Please familiarize yourself with the College's academic integrity policies. Be aware that I am contractually obligated to report students if I suspect that they have engaged in academic dishonesty. Lastly, unless otherwise directed, cell phones should be silenced and out of sight during all class periods.

Accessible Education Services (AES):

(AES) is located in the Goode-Pasfield Center for Learning and Teaching in Fintel Library. AES provides reasonable accommodations to students with documented disabilities. To register for services, students must self-identify to AES, complete the registration process, and provide current documentation of a disability along with recommendations from the qualified specialist. Please contact Laura Leonard, Assistant Director of Academic Services for Accessible Education, at 540-375-2247 or by e-mail at aes@roanoke.edu to schedule an appointment. If you have registered with AES in the past and would like to receive academic accommodations for this semester, please contact Laura Leonard at your earliest convenience to schedule an appointment. If you are on record with AES as having academic or physical needs requiring accommodations, please contact me as soon as possible. We need to discuss your accommodations before they can be implemented. Also, please note that arrangements for extended time on exams must be made at least one week before every exam.

<u>#</u>	<u>Date</u>	<u>Topic</u>	<u>Reading</u>	<u>Due</u>
		UNIT 1: Classical thermodynamics		
1	Aug. 19	Introduction and preview	--	
2	21	Heat, temperature and the 1 st Law	1.1-1.4	
3	24	Adiabatic processes	1.5-1.7	
4	26	Multiplicity and the 2 nd Law	2.1-2.3	
5	28	Entropy I	2.4-2.8	
6	31	The Carnot cycle	3.1-3.4	
7	Sep. 2	Reversibility and real engines	3.5-3.7	
8	4	Problem set workshop		
		UNIT 2: Transport theory		
9	7	Random walks	15.1-15.2	PS 1
10	9	Momentum transport and viscosity	15.3-15.4	
11	11	Thermal transport and diffusion	15.5-15.6	
12	14	Climate modeling I	6.4, Handouts	
13	16	Climate modeling II	Handouts	PS 2
14	18	Review and catch-up		
15	21	TEST 1		
		UNIT 3: Statistical mechanics		
16	23	Density of states	4.1	
17	25	General definition of temperature	4.2-4.4	Paper draft
18	28	Thermal probabilities	5.1-5.3	
19	30	The partition function	5.4-5.5	
20	Oct. 2	The canonical distribution	5.6-5.8	Paper final, PS 3
21	5	Chemical potential I	7.1	
22	7	Chemical potential II	7.2-7.5	
23	9	Ideal gas: quantum treatment	8.1-8.3	
24	12	Ideal gas: classical limits	8.4	
25	14	Free energy	10.1-10.3	
26	16	“Minimize the free energy”	10.4-10.8	
27	19	Chemical equilibrium	11.1-11.2	
28	21	Chemical equilibrium II	11.3	
29	23	Classical stat mech	13.1-13.2	PS 4
30	26	Equipartition theorem	13.3	
31	28	Review and catch-up		
32	30	TEST 2		

UNIT 4: Phase Transitions and the 3 rd Law				
33	Nov. 2	Phases of matter I	12.1-12.3	
34	4	Phases of matter II	12.4-12.5	
35	6	Fermions at low temperature I	9.1	
36	9	Fermions at low temperature II	9.1	
37	11	Prelude to the 3 rd Law	14.1-14.3	PS 5
38	13	The 3 rd Law	14.4-14.5	
39	16	Review and catch-up		
		FINAL: Saturday, November 21, 1:00-5:00 pm		

Note: You should expect to spend a combined total of 12 hours per week on lecture, homework, and reading for PHYS 370.